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PROCESS FOR THE DETERMINATION OF MTBE IN THE GROUND AND  
AIR.

The present invention relates to a process for the determination of pollution by methyl ter butyl ether.

5       Methyl ter butyl ether (MTBE) is the most widely used among oxygenated additives for motor vehicles. Its addition improves combustion and significantly reduces the emission of carbon monoxide, especially during low winter temperatures. The possibility of a  
10 leakage in the earth of fuels contained in underground tanks of service stations is probable. As a result of this, MTBE has been the object of a great deal of research with respect to its destiny in the environment and its potential impact on public health, mainly  
15 bearing in mind that this substance is extremely volatile and soluble in water. In addition, if present, it remains in deep water and sediments as owing to its very limited biodegradability, with an odour that can be noticed starting from concentrations at a level of

20 ppb. Its cancerogenous activity, if existing, seems to be small.

There are various methods for determining and measuring MTBE: they range from gas chromatography to  
5 IRA and flame-ionization, but they are all difficult to apply to the ground.

We have now overcome these problems by means of a process which allows the continuous monitoring of MTBE, in the ground and on the surface, using sensors in the  
10 solid state.

In accordance with this, the present invention relates to a process for monitoring methyl ter butyl ether (MTBE) vapours, in concentrations equal to or higher than 0.1 ppm, in the ground and overlying  
15 atmosphere comprising:

- a) adopting a series of MTBE vapour sensors of which at least one in the earth, equipped with a membrane permeable to gases and impermeable to water, and at least one in the air on the surface of the ground,  
20 these sensors consisting of
- a sensitive element made of a semi-conductor metal oxide containing platinum;
  - a heater capable of bringing the temperature of said sensitive element to a range of 300 and 500°C;
- 25 b) - continuously observing the resistance variations

of the sensitive elements by interaction with MTBE,  
- comparing the signals emitted by the sensor in the  
earth and the sensor in the air on the ground-surface;  
- evaluating on the basis of this comparison the  
5 presence and concentration of MTBE in the surface  
layers or depths of the ground and in the atmosphere  
above the ground itself.

A further object of the present invention relates  
to the device for effecting the process.

10 A typical embodiment of the invention is described  
hereunder, with reference to figures 1 and 2 in which  
equal numbers correspond to equal elements.

Figure 1 illustrates a sensor, in enlarged form.  
The sensitive element 1 is produced by placing by  
15 . screen printing, on an aluminum slab with dimen-  
sions of 3 x 9 x 0.25 mm, a 40 micron layer of a tin  
oxide paste, containing platinum. Powders are used  
having a particle size of less than 1 micron containing  
from 20 to 30% by weight of alumina and organometallic  
20 platinum as catalyst in a quantity ranging from 0.1 to  
1% by weight. A resistor (consisting of a layer of any  
commercial screen printing conductor paste capable of  
resisting at least 400°C) is deposited, again by  
screen printing, on the opposite side of the slab,  
25 to keep the sensitive element at an operating tempera-

ture of 300-500°C. After depositing the electric contacts also by screen printing, the slab is subjected to a baking step in an oven at 800-1000°C for an hour.

5 Finally the device, which forms the sensitive element, is assembled on a T078 2 container and inserted in a steel cylinder 3 closed by means of a flame-shield net 4. If the sensor described is fixed into the ground, a membrane 5, permeable to gases and imperme-  
10 able to water, is inserted under the flame-shield net to prevent any possible water present in the earth from entering into contact with the sensitive element. An appropriate porous septum or even better a membrane made of ePFTE material can be used for the purpose.

15 The sensitive elements can alternatively be produced with other types of semi-conductor metal oxides, but still using platinum as catalyst.

The sensors are equipped with feeders, or alternatively batteries, to supply energy to the heater and  
20 resistivity measurement circuit of the sensitive element.

Figure 2 illustrates an underground tank 6 of a service station for leadless fuels with a configuration with three sensors for the embodiment of the present  
25 invention. Two sensors 7, like those described with a

gas permeable membrane, are fixed in the ground at the sides of the tank, a sensor 8 without a membrane inserted in the chamber 9 above the tank. 10 illustrates the data acquisition switchboard.

5 Sensors such as those described above have a sensitivity which is such as to signal the presence of vapours of gasoline containing MTBE or MTBE alone with concentrations even less than 1 ppm in the air. The possibility of comparing the signals coming from the  
10 sensors fixed in the ground with those situated in the chamber above the tank over a period of time, make it possible to distinguish between leakages on the ground surface and losses from the underground tank.

In another embodiment of the same invention,  
15 sensors can be placed along an underground pipe around it and on the ground surface above. In this case the signals emitted from the sensors can be sent via radio to a central unit for collection and processing.

A few examples are provided below for a better  
20 understanding of the present invention but should not be considered as limiting the scope of the invention itself.

#### EXAMPLE 1

Using a sensor according to the one described  
25 above, and a tin container, conductivity measurements

are carried out in the presence of gasoline vapours to which 10% of MTBE has been added.

In figure 3.(a) the trace shows the kinetics response of the sensor in relation to the time at various concentrations of gasoline.

Figure 3.(b) shows the variation in the resistance in relation to the concentrations of gasoline. As can be seen the response is proportional to the concentration logarithm and allows concentrations of less than 1 ppm to be detected.

#### EXAMPLE 2

With the procedure described above, a system consisting of two MTBE sensors and an electronic control unit is prepared.

One of the sensors, protected by an ePFTE membrane, is inserted, up to a depth of about 10 cm, in a tank of 50 x 40 x 30 cm full of sandy earth. The second sensor is placed at about 20 cm from the first and about 5 cm from the surface.

After a stabilization period of about 30 minutes 1 ml of gasoline containing 10% of MTBE is injected with a syringe into the ground, at a distance of 10 cm from the underground sensor and at a depth of 10 cm. In figure 4 the temporal point of the injection is indicated with the arrow A. As can be observed, the trace

registered by the sensor in the air (2) indicates an almost immediate decrease in resistance, whereas the trace registered by the sensor in the ground (1) indicates a delay of about 5 min. before the decrease  
5 in resistance.

After a few hours, 1 ml of gasoline (indicated with the arrow B in figure 4B) is injected again. As can be observed, the trace of the sensor in the ground (1) begins to indicate a decrease in resistance start-  
10 ing from the level reached with the previous injection of gasoline. This shows that the sensor is capable of monitoring a further leakage also starting from ground which has already been polluted.

With respect to the trace in air, this starts from  
15 a much higher resistance value of the sensor as gasoline vapours dilute very rapidly in air, unlike the ground where the vapours interstitial tend to remain trapped.

The time delay of a few minutes in this case, shown by the response of the sensor fixed in the ground  
20 with respect to that in the air depends on the fact that in the earth interstitial vapours of gasoline and MTBE, although being mobile enough to allow this type of measurement, need a certain amount of time to spread from the leakage point to the sensor. In air the  
25 vapours obviously spread at a much faster rate and the

sensor consequently does not show significant delays.

As mentioned in the description, the different behaviour of sensors in the ground and in the air enables a leakage in the surface to be distinguished

5 from a leakage in depth in the ground.

Year	1990	1991	1992	1993	1994	1995	1996	1997	1998	1999	2000	2001	2002	2003	2004	2005	2006	2007	2008	2009	2010	2011	2012	2013	2014	2015	2016	2017	2018	2019	2020	2021	2022	2023	2024	2025	2026	2027	2028	2029	2030	2031	2032	2033	2034	2035	2036	2037	2038	2039	2040	2041	2042	2043	2044	2045	2046	2047	2048	2049	2050	2051	2052	2053	2054	2055	2056	2057	2058	2059	2060	2061	2062	2063	2064	2065	2066	2067	2068	2069	2070	2071	2072	2073	2074	2075	2076	2077	2078	2079	2080	2081	2082	2083	2084	2085	2086	2087	2088	2089	2090	2091	2092	2093	2094	2095	2096	2097	2098	2099	2100
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